Copper(II) coordination polymers with dimethyl(methylenoxyaryl) oxides-based ligands: Synthesis, structure and properties

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The coordination polymers are polymers whose repeating units are metal complexes. Recently, research in the field of design and synthesis of novel coordination polymers with predictable properties received particular attention due to their application as novel materials in various areas of industry. The ligands used for the synthesis have a polydentate nature and their characteristics such as nucleophilisity of the donors, bond angles, ligand length, bulkness, etc. play a crucial role in constitution of the supramolecular compounds. In addition, tendency of metal ions to adopt certain geometries also influences of the coordination polymer's structure and properties.

In the paper presented here, we describe the synthesis, single-crystal structure and physicochemical properties of three novel one dimensional copper(II) coordination polymers obtained with bis(dimethylphosphinylmethylenoxy)benzenes [1], Fig.1.



Fig. 1. Copper(II) coordination polymer of 1,3-bis(dimethylphosphinylmethylenoxy)benzene.

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References

1. S. Varbanov, T. Tosheva, G. Borisov; *Phosphourus, Sulfur & Silicon* 63 (1991) 397-402.